

1 **Effect of Temperature, Physical Aging, and Moisture Absorption**
2 **on Positron Lifetime Parameters in Cured Polyester Resin**
3 **Studied by Positron Annihilation Lifetime Spectroscopy**

4 **Abstract**

5
6 The positron annihilation method has been widely applied to study an extensive range of
7 materials such as metals, alloys, molecular solids, and numerous other technologically
8 important systems. Among the different experimental methods, positron annihilation lifetime
9 spectroscopy (PALS) has proven to be especially effective in polymer studies due to its high
10 sensitivity to the free-volume environments within the polymer. Examining these free-volume
11 features provides crucial information regarding the physical, mechanical, and thermal
12 responses of polymeric materials. In this study, PALS experiments were conducted on a cured
13 polyester resin to explore the responses of positron lifetime parameters to temperature
14 variations, physical aging, and moisture uptake. The lifetime spectra were processed using
15 the PATFIT analysis software. The third lifetime component (τ_3) and its corresponding
16 intensity (I_3), associated with the size and density of free-volume sites showed clear
17 variations under these different conditions. The observed trends are interpreted in relation to
18 modification in the free-volume structure within the polymer network.

19
20 Keywords: Positron annihilation, Free-volume, Glassy Polymers, ortho-Positronium lifetime,
21 Glass transition temperature, Physical aging

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28 *Correspondence Author *email:* haldarbidyut198@gmail.com

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30 **1. Introduction**

31 The interplay between atomic mobility and free volume in polymers and related
32 materials has long been a subject of scientific interest. The theoretical framework of free
33 volume, used to interpret molecular dynamics and the corresponding physical responses of
34 materials in their glassy and liquid states, was first established by Fox & Flory (1951,1950),

35 expanded by Doolittle (1951), and further developed by Liu et al. (1993). Since then, a
36 central objective in polymer research has been to quantify the relationship between molecular
37 dynamics and free-volume (FV) characteristics. At first, the concept of free volume (FV) was
38 viewed as mostly theoretical because it seemed impossible to measure it directly. The
39 challenge stemmed from the fact that these voids exist on the scale of only a few angstroms
40 and persist for extremely short time, around 10^{-13} seconds or more. Standard characterization
41 methods, including STM, SEM, X-ray diffraction, neutron diffraction, and AFM, are limited
42 to observing static surfaces or voids greater than roughly 10 Å in size. As a result, these
43 techniques are unable to detect the fleeting, sub-nanometre free-volume cavities that are
44 characteristic of polymeric materials.

45 Positron annihilation spectroscopy (PAS) has become a powerful technique for
46 probing free-volume hole (FVH) characteristics in polymers, as noted by Liu et al. (1993). In
47 this method, first demonstrated by Jean (1990) and Schrader & Jean (1988), positrons are
48 injected into the material, and their annihilation lifetimes both in free or trapped states,
49 including positronium (Ps), are recorded. Because positrons and Ps carry a positive charge,
50 they are naturally repelled by polymer ion cores and are therefore inclined to migrate toward
51 microscopic free-volume cavities that correspond to regions of lower electron density. Ferrell
52 (1956) explained that electron-exchange repulsion between Ps and neighbouring molecules
53 further promotes its localization in these open-volume sites. The annihilation photons largely
54 originate from these regions, enabling direct insight into the free-volume architecture.
55 Comparable observations have been documented in earlier and later investigations conducted
56 by Jean (1990, 1994), Haldar and co-workers (1996), Wang and colleagues (1990), and Yang
57 and co-authors (2015), who reported systematic variations in positron lifetimes with changes
58 in temperature, pressure, and physical aging.

59 When a positron enters a material, it may annihilate immediately with an electron or
60 may first form a transient bound state prior to annihilation. If no intermediate state is formed,
61 termed as direct annihilation, generally produces lifetimes on the order of 100–500
62 picoseconds. In contrast, in molecular liquids and solids, positrons commonly pair with
63 electrons to create a short-lived bound entity known as ‘*positronium*’ (Ps).

64 As outlined by Castelli (2012), positronium can exist in two possible spin alignments,
65 the singlet configuration, known as ‘*para-positronium*’ (p-Ps, 1S_0), where the electron and
66 positron possess opposite spins ($\downarrow\uparrow$), and the triplet state, termed ‘*ortho-positronium*’ (o-Ps,
67 3S_1), in which the spins are parallel ($\uparrow\uparrow$). Tao (1972) reported that p-Ps typically annihilates
68 into two photons with an average lifetime of roughly 125–130 picoseconds, whereas o-Ps

69 survives much longer, with a characteristic three-photon decay lifetime of about 1.47×10^{-7}
70 seconds. Within condensed phases, interactions with surrounding molecules can induce the
71 spin-state transition of ortho-positronium to its para state. This conversion pathway results in
72 a quicker two-gamma annihilation event and is broadly referred to as *quenching*. According
73 to Kato et al. (2020), such quenching mechanisms shorten the effective o-Ps pick-off lifetime,
74 which generally falls in the range of 1 to 10 nanoseconds.

75 According to Lue et al. (2008), the extremely small dimension of the positronium (Ps)
76 species, approximately 1.59 Å, makes positron annihilation spectroscopy (PAS) highly
77 effective for identifying free-volume holes (F VH) on the angstrom scale, as well as molecular
78 motions occurring on timescales of 10^{-10} s or longer. In contrast to many traditional
79 characterization methods, PAS can explore these minute cavities with minimal influence from
80 the surrounding bulk material. Studies by Consolati et al. (2023) show that, in molecular
81 materials, Ps is preferentially created within free-volume regions, where the long-lived o-Ps
82 component is particularly valuable because its lifetime correlates directly with the size of the
83 adjacent void. When o-Ps becomes trapped in an F VH, its observed lifetime acts as a highly
84 sensitive measure of the hole dimensions (Consolati et al., 2023).

85 Positron annihilation spectroscopy (PAS), and particularly its lifetime measurement
86 (PALS), has become a widely recognized quantitative tool for examining polymer structures.
87 As noted by Consolati et al. (2023), beyond determining the dimension and proportion of
88 F VH, the method can also provide insight into how these voids are distributed on the scale of
89 1–10 Å. This investigation uses PALS for the analysis of the dimensions and concentration of
90 F VH in cured polyester resin systems subjected to various experimental conditions.

91 Earlier investigations by Consolati et al. (2023) and Kaushik (2011) indicate that
92 cured epoxy and polyester resins fall under the category of glassy polymers, distinguished by
93 glass transition temperatures (T_g) that lie above ambient conditions. Understanding T_g is
94 essential for their practical use, since polymer mechanical performance declines markedly
95 once this limit is exceeded, resulting in diminished toughness and structural integrity, as
96 reported by Yang et al. (2015) and Odegard & Bandyapadhyay (2011). The response of
97 glassy polymers is strongly influenced by their molecular packing efficiency, often described
98 in terms of free volume (FV). Below T_g , the available FV remains largely immobile and
99 molecular motion is minimal. As the material nears T_g , the FV grows to a level that permits
100 increased segmental mobility. Once the temperature surpasses T_g , the polymer transitions

101 into a viscoelastic regime, during which FV rises rapidly with temperature until it nears the
102 melting region.

103 Significant research attention is being directed toward understanding physical aging
104 processes in polymeric materials. The occurrence of physical aging in polymers has been
105 reported for more than two decades but it is only recently that this subject has become the
106 focus of widespread study (Wang et al., 2003; Kaushik, 2011; Odegard & Bandyapadhyay,
107 2011; Merrick et al., 2020). The major issues associated with these polymers is their
108 degradation over extended periods, which can result from the breaking of chemical bonds or
109 alterations in their internal microstructure. Glassy polymers are commonly described as
110 supercooled liquids that have transitioned into a solid, non-equilibrium condition. Over time,
111 they experience a gradual structural reorganization, known as physical aging, as the material
112 slowly approaches thermodynamic equilibrium (Odegard and Bandyapadhyay, 2011; Gordo
113 et al., 2013). This aging phenomenon is generally accompanied by changes in free-volume
114 characteristics, such as a decrease in void dimensions and a rearrangement of free-volume
115 sites.

116 The situation is most often demonstrated using the volume vs. time curve that is
117 observed for amorphous polymers as they are cooled from a completely melted condition
118 through the glass transition temperature, T_g , to some lower temperature (Hill, 1999). In
119 amorphous polymers and amorphous phases of semicrystalline materials, rapid cooling below
120 the glass transition temperature (T_g) traps polymer chains in non-equilibrium states, leading
121 to excess volume that depends on the cooling rate (Hill, 1999). Accordingly, a systematic
122 study of polymer aging was carried out under varying conditions.

123 Moisture uptake is another critical parameter influencing the mechanical behaviour of
124 glassy polymers. Water molecules slowly penetrate the polymer network, where they reside
125 in free-volume cavities and function as plasticizing agents (Gordo et al., 2013; Lue et al,
126 2008; Wang et al., 2003). This interaction leads to material softening, resulting in a decline in
127 mechanical stability and performance. The present work examines the impact of moisture
128 absorption under varying environmental conditions on the free-volume properties of these
129 polymers.

130 Although numerous PALS investigations have addressed the effects of temperature,
131 physical aging, and moisture absorption in cured epoxy systems (Davis and Pethrick, 1998;
132 Wang et al., 2003; Yang et al., 2015; Gordo et al., 2013), similar investigations focusing on
133 cured polyester polymers are relatively scarce (Kaushik, 2011). In this work, a detailed PALS

134 evaluation of cured polyester resins has been performed to bridge the existing gap. offering a
135 quantitative evaluation of free-volume evolution under varying thermal conditions, aging
136 processes, and moisture exposure.

137 In positron annihilation lifetime spectroscopy (PALS), the experimentally measured
138 positron lifetime (τ) is influenced by the extent of spatial overlap between the positron (ρ^+)
139 and electron (ρ^-) density distributions at the annihilation site (Schrader and Jean, 1988; Jean,
140 1993). To describe positronium localization in condensed matter, Tao (1972) proposed a
141 quantum mechanical model in which positronium (Ps) is confined within a spherical potential
142 well of radius R_0 featuring infinitely large potential barriers. On the basis of this model, a
143 widely adopted semi-empirical relation was subsequently formulated, also known as Tao–
144 Eldrup model, establishing a direct correlation between the o-Ps lifetime (τ_3) and the radius
145 (R) of FV cavities distributed within the polymer matrix (Mills, 1981; Nakanishi et al., 1988).

$$\tau_3 = 0.5 \left[1 - \frac{R}{R_0} + \frac{\sin 2\pi(R/R_0)}{2\pi} \right]^{-1} \quad (1)$$

146 where $R_0 = R + 1.66$ is in Å and τ_3 is in ns (10^{-9} sec).

147 A thorough analysis of the positron annihilation lifetime (PALS) spectrum is crucial to
148 ensure meaningful physical interpretation. Quantitative interpretation of PAL spectra is
149 typically achieved through finite-component lifetime analysis. In this method, the
150 experimental spectra are decomposed using a least-squares fitting routine to extract the
151 annihilation rates, $\lambda_i = 1/\tau_i$ ($i = 1, 2, 3, 4$), together with their associated intensities I_i . Such
152 analyses are commonly carried out using the PATFIT computational package (Kirkegaard *et*
153 *al.*, 1989). In the present work, the PAL spectra were resolved into three separate lifetime
154 components. The shortest component ($\tau_1 \approx 0.12$ ns) is associated with the intrinsic
155 annihilation of singlet para-positronium (p-Ps). The lifetime of intermediate duration
156 ($\tau_2 \approx 0.40$ ns) corresponds to free positron annihilation with electrons in the bulk matrix,
157 while the longest lifetime ($\tau_3 \geq 0.5$ ns) arises from pick-off annihilation of ortho-positronium
158 (o-Ps) localized within FV cavities. The parameter τ_3 is subsequently employed to determine
159 the mean free-volume hole size using Eq. (1), whereas its intensity I_3 provides information on
160 the relative population and spatial distribution of these cavities. Moreover, free-volume
161 fraction (f_v) of the polymer may be estimated using a semi-empirical relation proposed by
162 Wang *et al.* (1990) and Jean (1994):

$$f_v = A \times V_t \times I_3 \quad (2)$$

164 where V_f denotes the free-volume hole size (in \AA^3) derived from τ_3 , and A is a proportionality
165 constant representing the o-Ps formation probability. The value of A, typically in the range
166 0.001–0.002, is obtained from the polymer's volumetric thermal expansion coefficient
167

168 **2. Experimental Methodology**

169 **2.1 Sample preparation**

170 Polyester specimens were prepared by curing unsaturated styrenated alkyd resins
171 using methyl ethyl ketone peroxide (MEKP, 1 wt%) as the initiator and cobalt octoate (0.5
172 wt%) as the accelerator. Curing was performed at 70 ^0C for 12 h, followed by post-curing at
173 129 ^0C for 4 h and 140 ^0C for 2 h to ensure complete crosslinking. The cured laminates were
174 cut into 1 cm \times 1 cm samples for positron annihilation measurements.
175

176 **2.2 Temperature-dependence of Positron annihilation lifetime measurements**

177 Positron annihilation lifetime (PAL) spectra were recorded at 25 ^0C using a standard
178 fast–fast coincidence spectrometer setup equipped with a ^{22}Na positron source of 15 μCi
179 activity. The spectrometer's time resolution, evaluated from the prompt spectrum of a ^{60}Co
180 source, was 0.32 ns (FWHM). Temperature-dependent measurements were carried out by
181 mounting the source–sample assembly in a custom-designed heating cell controlled by a
182 microprocessor-based temperature controller (Indotherm MPC 500) with an accuracy of ± 1
183 K. Lifetime spectra were collected over the temperature range 20–200 ^0C at intervals of
184 approximately 10 ^0C . Each spectrum contained at least 5×10^6 counts to ensure adequate
185 statistical precision. The spectra were analysed using the PATFIT program and resolved into
186 three lifetime components.
187

188 **2.3 Physical aging experiments**

189 To investigate physical aging effects, cured polyester samples were annealed under
190 vacuum at 150 ^0C for 100 h. Following annealing, the samples were cooled using different
191 cooling rates, namely furnace cooling, water quenching, and liquid-nitrogen quenching. PAL
192 measurements were subsequently conducted under room-temperature conditions at regular
193 time intervals to monitor aging-induced changes.
194

195 **2.4 Moisture absorption studies**

196 Moisture uptake experiments were conducted using a specially designed glass apparatus
197 consisting of two interconnected chambers, one containing the polymer specimen and the

198 other filled with water. Both chambers were evacuated independently prior to opening the
199 connecting valve, allowing water vapor to diffuse into the sample chamber. Additional
200 absorption studies were performed by directly immersing the samples in water at 20 $^{\circ}\text{C}$ and
201 97 $^{\circ}\text{C}$. Positron lifetime spectra were recorded after exposure times of 15 min, 30 min, 1 h,
202 2 h, 4 h, 8 h, and 16 h under each condition.

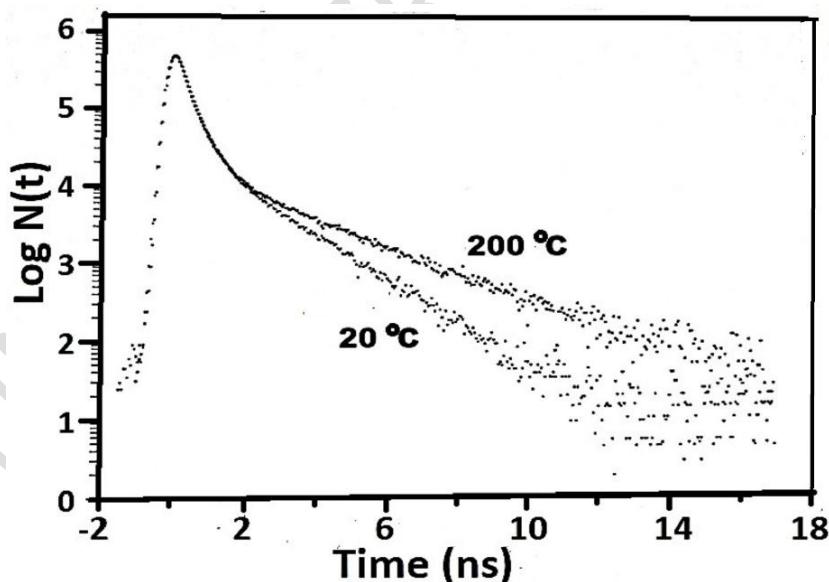
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204 **3. Results and Discussions**

205 **3.1 Temperature-dependence of positron lifetime parameters**

206 Fig.1. compares the positron lifetime spectra of polyester specimens measured at 20
207 $^{\circ}\text{C}$ and 200 $^{\circ}\text{C}$, revealing clear temperature-induced changes in spectral shape. The evolution
208 of the ortho-positronium lifetime (τ_3) and its relative intensity (I_3) with temperature is
209 depicted in Fig. 2. Since both parameters are closely linked to the free-volume structure of
210 polymers, the analysis that follows primarily examines their temperature-dependent
211 behaviour. Under the assumption of spherical free-volume cavities, the semi-empirical
212 expression given in Eq. (1) was used to derive the average hole radius (R) from the measured
213 τ_3 values. The corresponding FVH size was subsequently calculated as $V_f = \frac{4}{3}\pi R^3$, with
214 these values displayed on the secondary axis of Fig. 2(a).

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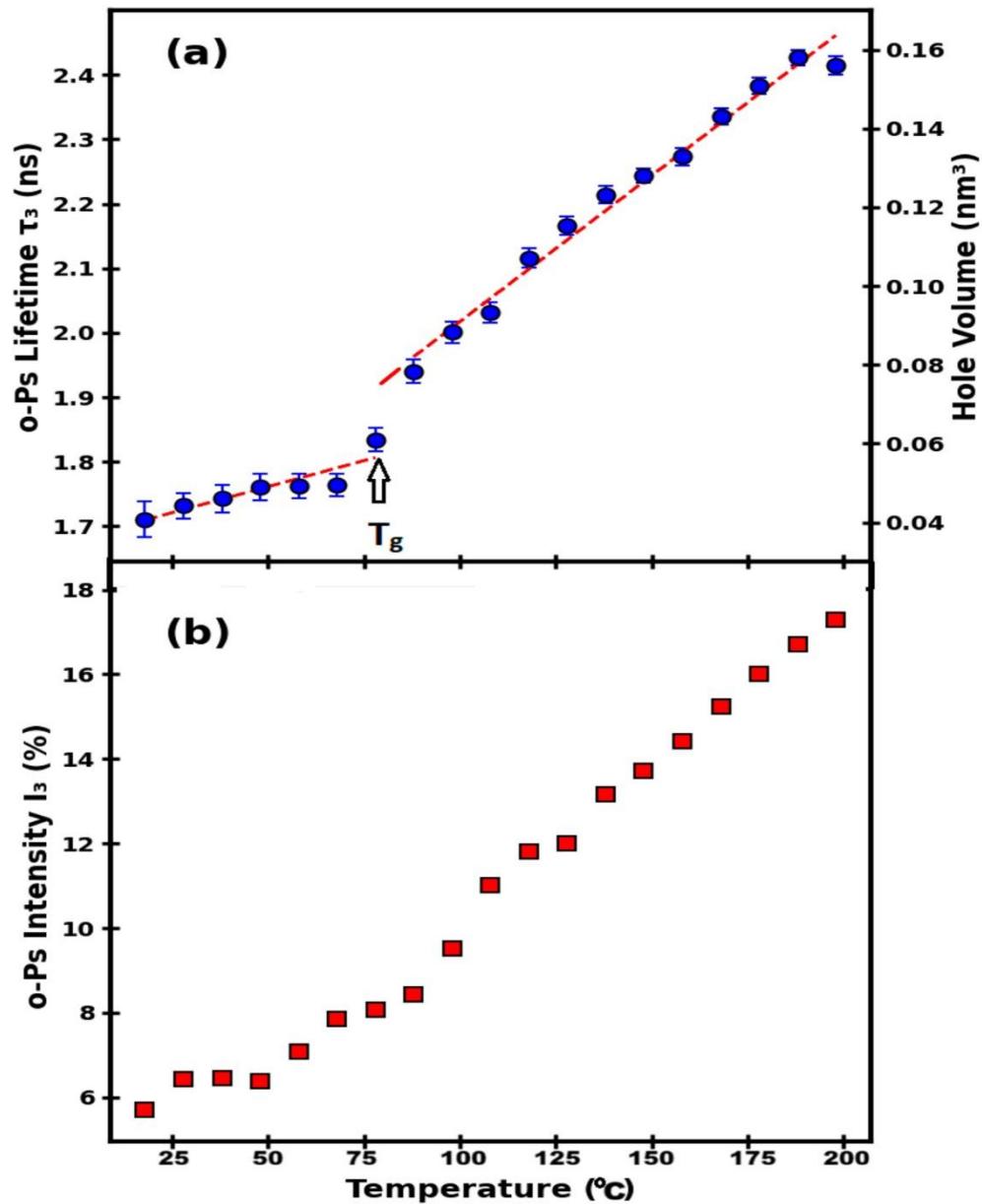


216

217 **Figure 1.**PALS spectra of the polyester sample recorded at two temperatures.
218

219 The variation of τ_3 with temperature, shown in Fig. 2(a), reveals two well-defined regions
220 separated by a marked change in slope near 77 $^{\circ}\text{C}$. Above this temperature, τ_3 increases at a
221 significantly faster rate, suggesting the onset of the glass transition, identified here as $T_g = 77$

222 ^0C . In the lower-temperature regime, the material exhibits typical glassy behaviour, whereas
 223 at temperatures exceeding T_g , the polymer exhibits a transition from the glassy to the rubbery
 224 state.

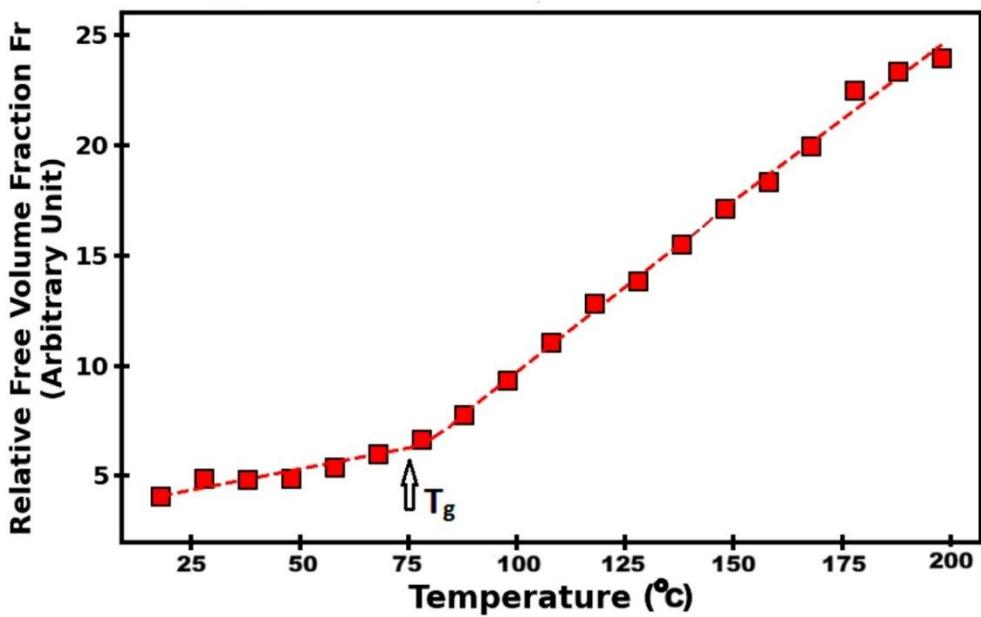


225
 226 **Figure 2.** Temperature-dependent behaviour of (a) τ_3 (o-Ps lifetimes) and
 227 (b) I_3 (o-Ps intensity) in polyester sample
 228

229 At ambient temperature, the intensity I_3 of the polyester sample is approximately 7%,
 230 which is substantially lower than the values commonly reported for epoxy systems (~20%)
 231 (Jean et al., 1986; Deng and Jean, 1993). As the temperature is raised, I_3 increases
 232 monotonically, attaining a value of nearly 17% at 200 ^0C . This trend indicates an approximate
 233 2.4-fold enhancement in the population of free-volume cavities. Such temperature sensitivity
 234 of I_3 stands in contrast to epoxy polymers, where I_3 remains largely invariant with

235 temperature (Deng and Jean, 1993). Consequently, the overall growth in free volume
236 observed in the present system arises from the combined effects of cavity expansion,
237 reflected in τ_3 , and an increase in cavity concentration, as indicated by I_3 .

238 The contrasting behaviour can be explained by differences in polymer morphology.
239 Epoxy systems are predominantly amorphous, whereas polyesters possess a semi-crystalline
240 structure comprising amorphous regions interspersed with partially ordered domains
241 (Valsange et al., 2024; Deng et al., 1992). Studies have shown that o-Ps formation occurs
242 exclusively within the amorphous regions of polymers. While the o-Ps lifetime (τ_3) remains
243 largely insensitive to crystallinity, its intensity (I_3) decreases systematically with increasing
244 crystalline content, reflecting a reduction in the number of available free-volume sites rather
245 than a change in their size. The linear dependence of I_3 on crystallinity confirms that
246 positronium does not probe crystalline domains and allows crystallinity to be independently
247 assessed. Thus, at room temperature, the higher level of chain packing and molecular
248 ordering in polyester limits the



249
250 **Figure 3.** Temperature dependence of the relative free-volume fraction $F_r (= I_3 V_t)$
251 in the polyester sample, derived from Fig. 2.
252

253 availability of free-volume sites, resulting in relatively smaller initial cavity sizes. As
254 temperature increases, thermal agitation progressively disrupts this ordered arrangement,
255 leading to the formation of additional free volume and a corresponding rise in the o-Ps
256 intensity I_3 . At sufficiently elevated temperatures, the polymer structure approaches a highly
257 disordered state.

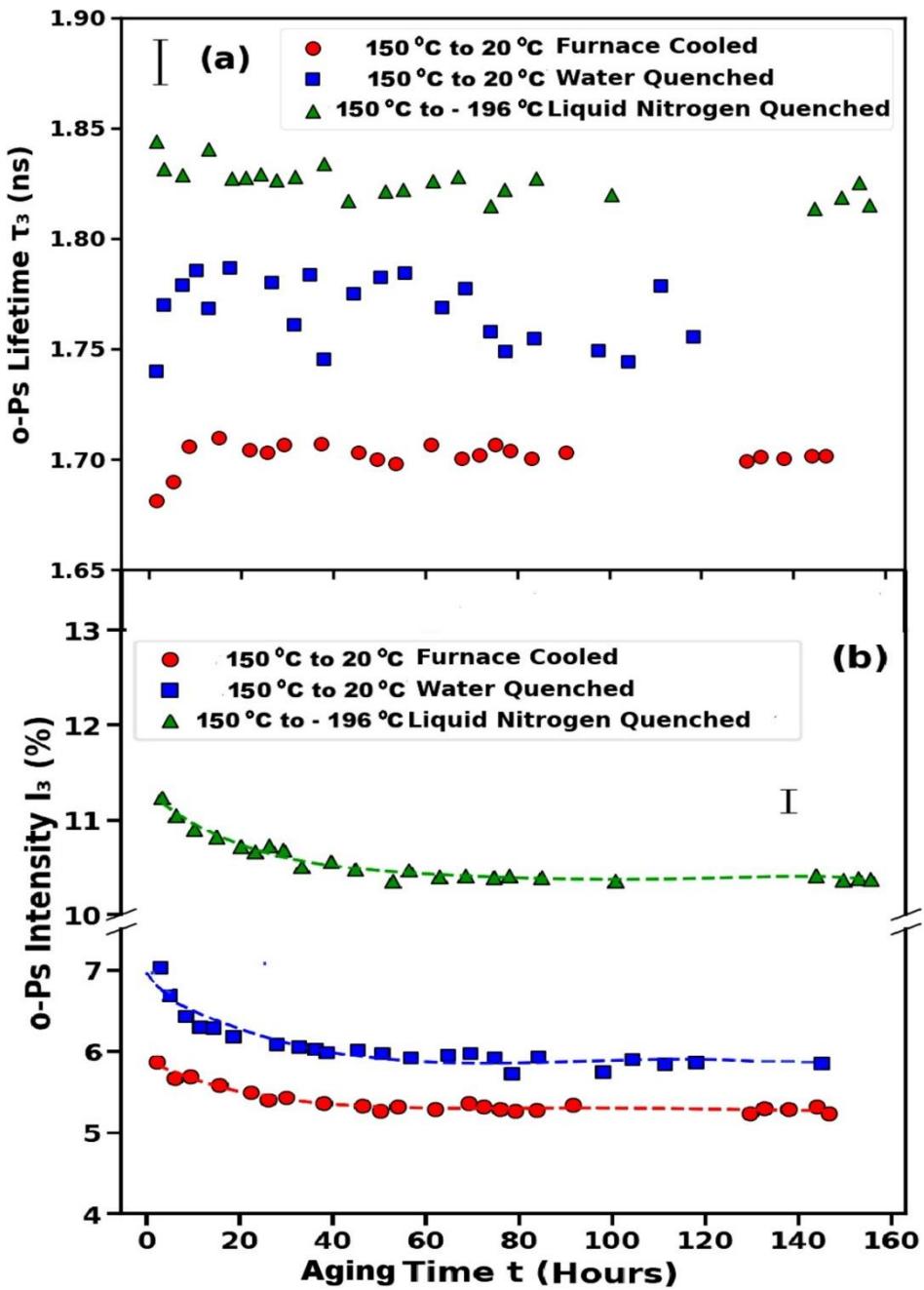
258 Considering the value of A as 1 in Eq. (2), the temperature dependence of the relative
259 free-volume fraction $F_r (= I_3 V_t)$, based on data of Fig.2, is shown in Fig. 3. A significant rise
260 in F_r is observed between 77 °C and 200 °C, arising from the simultaneous enlargement of
261 free-volume cavities and an increase in their number density.

262

263 **3.2 Study of structural relaxation and physical aging**

264 Physical aging in polymers refers to a reversible rearrangement of molecular packing
265 that occurs without the rupture or formation of chemical bonds. This phenomenon proceeds
266 over extended timescales and may require several decades to approach equilibrium (Consolati
267 et al., 2023). As reported by Consolati et al. (2023), when a polymer is thermally treated
268 beyond its glass transition temperature (T_g) and subsequently quenched to temperatures
269 below T_g , the free-volume configuration characteristic of the high-temperature state becomes
270 kinetically trapped i.e. the polymer retains a non-equilibrium free-volume state characteristic
271 of the higher temperature and hence a higher specific volume than its equilibrium value,
272 creating excess free volume. Once in the glassy regime, the system slowly evolves toward a
273 new thermodynamic equilibrium appropriate for the lower temperature (Struik, 1978). This
274 process is inherently sluggish due to the severely restricted segmental motion of polymer
275 chains within the glassy phase, where accessible free volume is scarce. Because the ortho-
276 positronium lifetime (τ_3) and its associated intensity (I_3) are highly sensitive to both the
277 dimension and population of FV cavities, the gradual structural relaxation accompanying
278 physical aging is expected to manifest as time-dependent variations in these positron
279 annihilation parameters (Hill, 1999; Cheng et al., 2009).

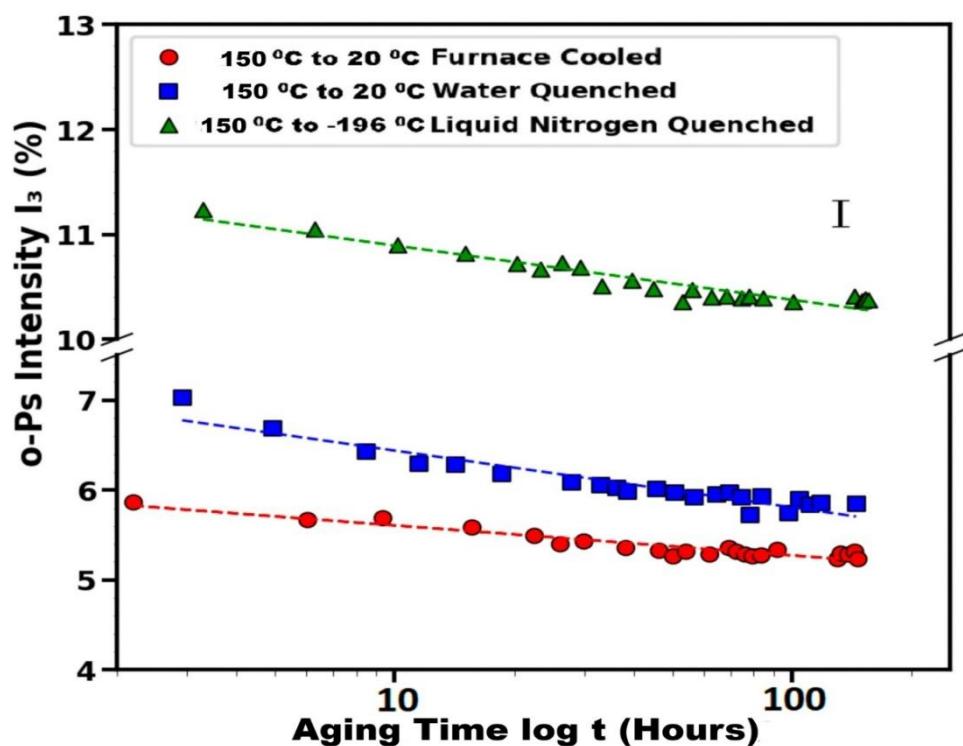
280 Fig. 4. presents the evolution of the o-positronium lifetime parameters of the polyester
281 sample plotted against aging time for three distinct cooling methods. Irrespective of the
282 cooling rate employed, τ_3 and I_3 exhibit comparable temporal trends, suggesting that the
283 qualitative features of physical aging are largely independent of the cooling history. Under
284 certain cooling conditions, the τ_3 values remain essentially invariant throughout the
285 observation period, whereas I_3 shows a slight but systematic decline with increasing aging
286 time. In addition, the decrease in I_3 becomes progressively less pronounced at longer aging
287 durations, indicating a deceleration of the aging kinetics. These observations imply that
288 physical aging predominantly



289
290 **Figure 4.** Time evolution of (a) τ_3 and (b) I_3 during physical aging of polyester
291 samples cooled at different rates.
292

293 influences the density of FV cavities, indicated by changes in I_3 , while their average size, as
294 reflected by τ_3 , is hardly affected. The gradual reduction in the aging rate further supports the
295 notion that the polymer undergoes a slow structural relaxation toward a new thermodynamic
296 equilibrium, a process expected to extend over prolonged timescales. Comparable aging
297 behaviour has been reported in earlier positron annihilation studies on polymer systems
298 (Hsu, Chai-Wen et al., 2015; Wang et al., 2003; Deng et al., 1992). It is observed that samples

299 subjected to higher cooling rates exhibit larger values of both τ_3 and I_3 at room temperature
 300 when compared with the as-prepared material. In particular, specimens quenched in liquid
 301 nitrogen display markedly higher o-Ps lifetimes and intensities than those cooled more slowly
 302 in a furnace or quenched in water. Such behaviour arises because rapid quenching suppresses
 303 structural relaxation during cooling, thereby trapping a higher level of molecular disorder
 304 within the polymer matrix. This frozen-in disorder enhances both the available free volume
 305 and the relative free-volume fraction. Fig.5. illustrates the dependence of o-Ps intensity (I_3)
 306 as a function of the logarithm of aging time. In all cooling conditions, I_3 exhibits an
 307 approximately linear dependence on $\log(t)$, indicating that the physical aging process follows
 308 an exponential decay with time. Such logarithmic time dependence is characteristic of
 309 diffusion- and relaxation-controlled processes in polymers (Struik, 1978).



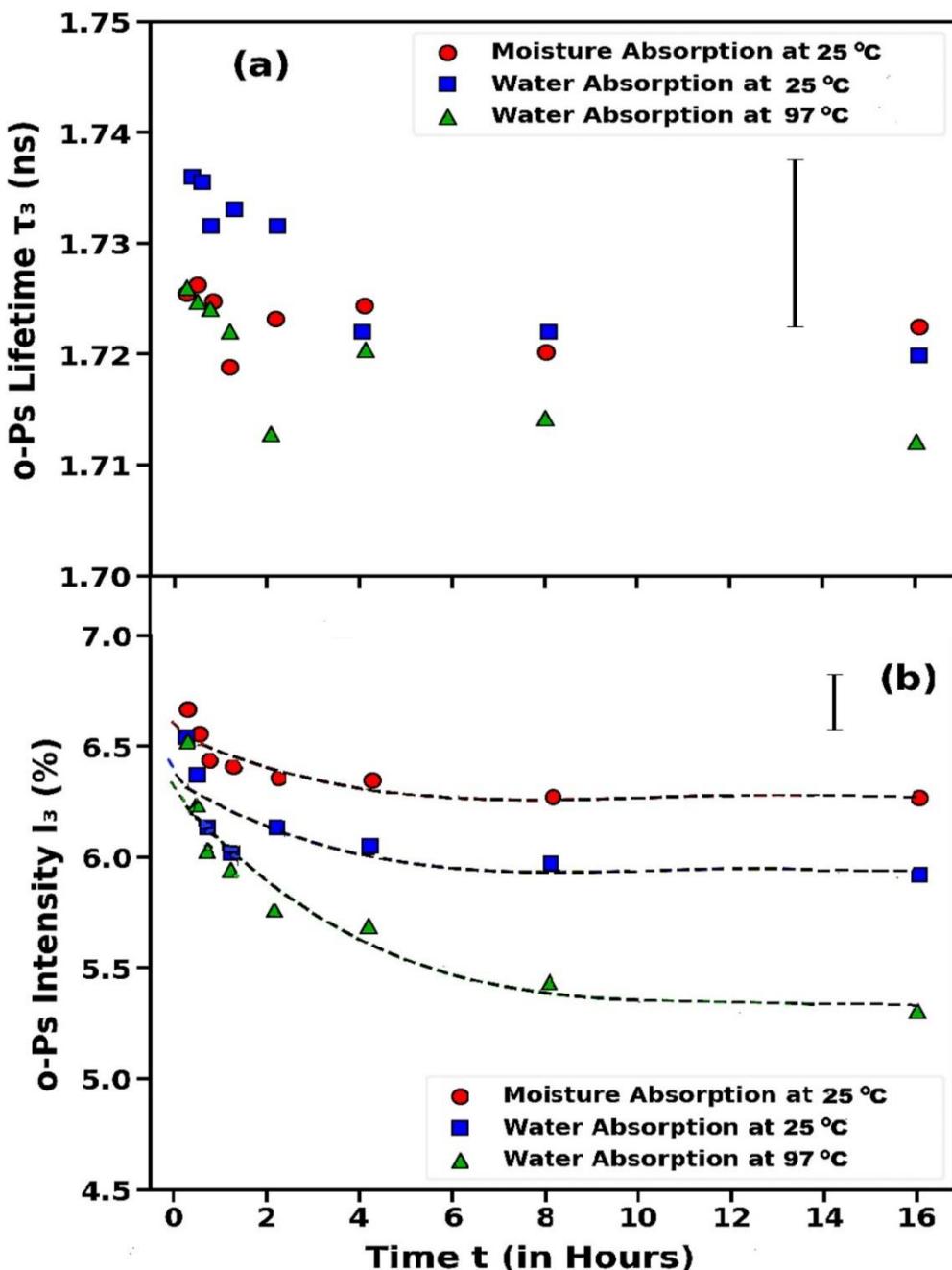
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311 **Figure 5.** Effect of physical aging in polyester samples cooled at different rates
 312 on I_3 as a function of $\log(t)$.
 313

314 **3.3 Water/moisture absorption studies**

315 Fig. 6. depicts the temporal evolution of the measured o-Ps lifetime values, τ_3 and
 316 I_3 for the polyester samples subjected to moisture exposure under different absorption
 317 conditions, as outlined earlier. The results demonstrate the impact of water uptake on the

318 polymer's free-volume characteristics of the polymer as a function of exposure time. While τ_3
319 remains



320
321 **Figure 6.** Influence of moisture/water absorption on the o-positronium lifetime
322 parameters of polyester: (a) τ_3 and (b) I_3 as a function of absorption time
323 under different exposure conditions
324

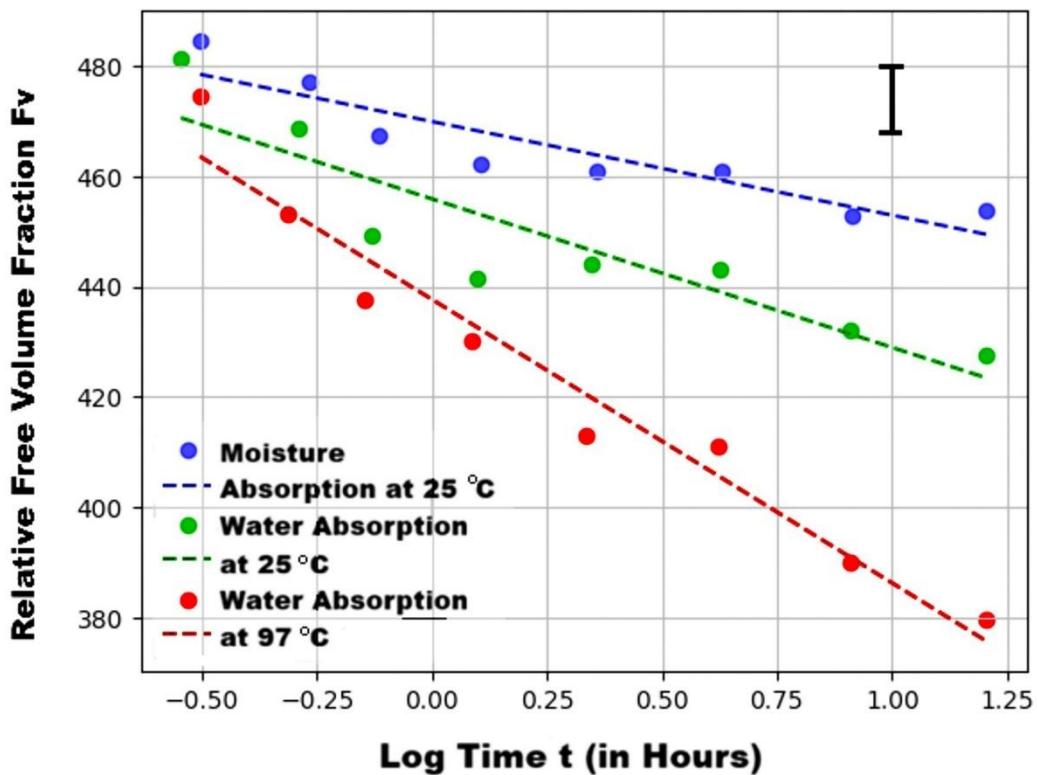
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326 essentially insensitive to the duration of moisture absorption across all samples, a progressive
327 reduction in I_3 is observed with increasing absorption time, followed by the attainment of a
328 steady-state value. The invariance of τ_3 indicates that moisture uptake does not appreciably

329 change the average size of the FV cavities, suggesting that water molecules preferentially
330 occupy pre-existing voids rather than inducing cavity expansion or polymer swelling within
331 the investigated time scale. The mean τ_3 value is approximately 1.72 ± 0.015 ns, which, based
332 on Eq. (1), corresponds to free-volume cavities with an effective spherical radius of about 2.5
333 ± 0.02 Å—comparable to the kinetic diameter of a water molecule (~ 3 Å). This dimensional
334 similarity suggests that water molecules introduced during moisture uptake can gradually
335 occupy existing free-volume sites within the polymer matrix, thereby limiting the number of
336 cavities available for positronium formation. Consequently, the observed decrease in I_3 with
337 absorption time can be attributed to the filling of intermolecular voids by water molecules
338 (Gordo et al., 2013), despite the inherently hydrophobic nature of polyester resins. Thus, the
339 reduction in I_3 therefore reflects a decrease in the number of positronium-forming sites rather
340 than a modification of cavity geometry, consistent with competitive occupation of FV holes
341 by water molecules. Furthermore, the magnitude of the o-Ps intensity (I_3) decrease is more
342 pronounced at elevated absorption temperatures, which may be associated with enhanced
343 free-volume availability and/or increased diffusivity of water molecules, enabling more rapid
344 and extensive penetration into the polymer network (Crank, 1975). Moreover, the attainment
345 of a plateau in I_3 at longer absorption times indicates saturation of accessible free-volume
346 sites, beyond which additional moisture uptake has a negligible effect on positronium
347 formation

348 Although polyester resins are generally considered hydrophobic, the presence of polar
349 ester groups and interfacial free-volume regions enables limited but measurable moisture
350 uptake. In contrast to physical aging, where changes are governed by intrinsic structural
351 relaxation, moisture absorption introduces an extrinsic mechanism that selectively reduces
352 the availability of free-volume sites without altering their characteristic size.

353 Fig. 7. presents the dependence of the o-positronium intensity (I_3) versus the
354 logarithm of moisture absorption time, derived from the data shown in Fig. 6(b). A similar
355 logarithmic time dependence of o-Ps intensity during moisture absorption has been reported
356 in several polymer systems and is generally associated with diffusion-limited occupation of
357 free-volume sites (Deng and Jean, 1993; Wang et al., 2003). The data points exhibit an
358 approximately linear dependence of I_3 on $\log(\text{time})$ over the investigated absorption period,
359 as indicated by the best-fit straight line. This behaviour suggests that the reduction in o-Ps
360 intensity follows a logarithmic time dependence, which is characteristic of diffusion-
361 controlled and relaxation-driven processes in polymer systems. The linear relationship
362 between I_3 and $\log(\text{time})$ implies that moisture uptake progressively reduces the number of

363 available



364

365 **Figure 7.** Dependence of o-positronium intensity (I_3) on the logarithm of
366 moisture absorption time for the polyester sample, corresponding to
367 Fig. 6(b), showing the best linear fit.

368

369 positronium-forming free-volume sites, with the rate of reduction slowing at longer times as
370 the system approaches saturation. Such kinetics are consistent with gradual occupation of
371 accessible free-volume cavities by water molecules, followed by a diminishing availability of
372 unfilled sites(Wang et al. 2003). The observed trend further supports the interpretation that
373 moisture absorption primarily influences the population of free-volume holes rather than their
374 average size, in agreement with the near-constant τ_3 values shown in Fig. 6(a).

375

376 **4. Conclusions**

377 Positron annihilation lifetime spectroscopy (PALS) was employed to investigate the
378 free volume in cured polyester polymers. The temperature dependence of the o-Ps lifetime
379 (τ_3) and its intensity (I_3) was analysed to determine the variation in free-volume hole size (V_f)
380 over the temperature range of 20–200 °C. Polyester polymers exhibited a pronounced
381 increase in V_f above the glass transition temperature (T_g). This behaviour was explored and
382 discussed in detail. Additionally, positron lifetime measurements during physical aging
383 revealed systematic changes over time that were consistent regardless of the cooling rate

384 applied to the samples. Finally, the effect of water absorption on the ortho-positronium
385 lifetime and intensity was examined, showing more significant changes when absorption
386 occurred at an elevated temperature of 97 $^{\circ}\text{C}$.

387

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