



ISSN NO. 2320-5407

Journal homepage: <http://www.journalijar.com>

INTERNATIONAL JOURNAL
OF ADVANCED RESEARCH

RESEARCH ARTICLE

Effect of culture media on biosynthesis of titanium dioxide nanoparticles using *Lactobacillus crispatus*

¹Jehan Abdul Sattar Salman, ²Kawther Hkeem Ibrahim, ³Fattma Abodi Ali

1,2. Department of Biology/College of Science/Al-Mustansiriya University/ Baghdad- Iraq

3. Department of microbiology/college of medicine/Hawler Medical University/Erbil-Iraq

Manuscript Info

Manuscript History:

Received: 23 March 2014
Final Accepted: 19 April 2014
Published Online: May 2014

Key words:

Biosynthesis, titanium dioxide nanoparticles, *Lactobacillus crispatus*, culture media

*Corresponding Author

Jehan Abdul Sattar

Abstract

The present study included the rapid synthesis of Titanium dioxide nanoparticles using *Lactobacillus crispatus* and focus on the effect of different culture media (MRS broth, JKH₁ broth and Whey solution) on TiO₂ nanoparticles synthesis. The synthesized nanoparticles were characterized using X-ray diffraction (XRD) technique, Atomic Force Microscopy (AFM) and Scanning electron microscopic (SEM). The XRD pattern of the sample showed the presence of main peak of $\theta=25.3$ matches the (101) crystallographic plane of anatase of TiO₂ nanoparticles. The morphological characteristics were found to be spherical, oval in shape, have the average size (70.98, 116.89, 113.67)nm synthesized in MRS broth, JKH₁ broth and whey solution respectively. Adding of 10% glucose to JKH₁ broth and Whey solution lead up to decreasing the average size of synthesized nanoparticles reached to (92.65 and 112.26)nm for JKH₁ broth and whey solution respectively.

Copy Right, IJAR, 2014. All rights reserved.

INTRODUCTION

Presently, there are chemical, physical and biological (including the use of microorganisms) routes available for the synthesis of metal oxide nanoparticles (Ahmad *et al.*, 2013). The chemical and physical synthesis of nanoparticles is expensive and often involves the use of toxic, hazardous chemicals which may pose environmental risks (Malarkodi *et al.*, 2013). Biological methods have been put ahead to be advantageous over other synthetic methods as they are cost effective and do not involve the use of toxic chemicals, high pressure, energy and temperatures (Tripathy *et al.*, 2010; Malarkodi *et al.*, 2013). The biological method for nanoparticle production is simple, eco-friendly and allows for getting controlled nanoparticles which can be used as catalysts with specific composition, which cannot be synthesized by classical methods, applications in sensors and medicine are envisaged and the nanoparticles synthesized in the bacteria can be used against the human pathogens (Popescu *et al.*, 2010). The microorganisms are used as possible "nanofactories" for development of clean, nontoxic and environmentally friendly methods for producing nanoparticles (Klaus-Joerger *et al.*, 2001). Nanoparticles are biosynthesized when the microorganisms grab target ions from their environment and then turn the metal ions into the elemental metal through enzymes generated by the cell activities (Li *et al.*, 2011). The microorganisms such as *Lactobacillus* sp. and *Saccharomyces cerevisiae* are used for the synthesis of titanium dioxide (TiO₂) nanoparticles (Jha *et al.*, 2009).

Biologically synthesized nanoparticles have wide application viz., biosensors, biolabelling, in cancer therapeutic and in coating of medical appliances (Prakash *et al.*, 2010). The oxide nanoparticles synthesized by several methods appear more and more useful, because these nanoparticles have good electrical, optical and magnetic properties that are different from their bulk counterparts (Sagadevan, 2013). The TiO₂ nanoparticles may be one of the most important materials for photocatalysts, cosmetics, and pharmaceuticals (Kirithi *et al.*, 2011). TiO₂, therefore is a

versatile material that has applications in various products such as paint pigments, sunscreen lotions, electrochemical electrodes, capacitors, solar cells, and even as a food coloring agent and in toothpastes (Byranvand *et al.*,2013). So, far only few reports are available on the biosynthesis of TiO₂ nanoparticles using *Lactobacillus* sp. Thus in this paper, we report on the synthesis of TiO₂ nanoparticles using *Lactobacillus crispatus* and the effect of different culture media on TiO₂ nanoparticles synthesis are also compared.

Material and Methods:

-*Lactobacillus crispatus*

Lactobacillus crispatus (isolated from vagina of Iraqi healthy women) have been selected as the best isolate among seventy five isolates of *Lactobacillus* spp. isolated and identified during another study (Data not show).

-Synthesis of titanium nanoparticles using *Lactobacillus crispatus*:

A-Using MRS broth

Three flasks were used, each flasks were filled with 40 ml of MRS(De-Man Rogosa Sharpe) broth . Then 20 ml of TiO₂(0.025m) were added to the first and second flask respectively and both were stirred for half hour on a magnetic stirrer while the third flask contain MRS broth only. Final concentration ultimately would be equivalent. *Lactobacillus crispatus* was cultured in first and third flask into CO₂-incubator at 37C° for (24,48,72) hours. Second flask was used as blank for first one , the change in color from light brown to dark brown observed and production of sediment will observed as primary detection of produced TiO₂ nanoparticles(Azhar *et al.*,2011).

B-Using JKH₁ broth

Three flasks were used, each flasks were filled with 40 ml of JKH₁(Jehan-Khawlah 1)broth which prepared from Gourd juice, the clarified juice was collected by coarse filtration (Whatman 40), then the pH was adjusted to 6.2 , autoclaved for 10 min.(Salman and khalaf , 2014). Then 20 ml of TiO₂(0.025m) were added to the first and second flask respectively and both were stirred for half hour on a magnetic stirrer while the third flask contain JKH₁ broth only. *Lactobacillus crispatus* was cultured in first and third flask into CO₂-incubator at 37C° for (24,48,72) hours. Second flask was used as blank for first one , the change in color observed and production of sediment was observed as primary detection of produced TiO₂ nanoparticles. For another experiment 10% glucose was added to JKH₁ broth and the same procedure for TiO₂ nanoparticles synthesis was done.

C-Using Whey solution

In a typical procedure of nanoparticles synthesis, *Lactobacillus crispatus* inoculated into sterilized 250 ml of whole milk in 500 ml flask and incubated for curdling at 37°C for 24 hours. The whey was collected by coarse filtration (Whatman 40). The filtrate was pale yellow in appearance(Ranganath *et al.*,2012).Three flasks were filled with 40 ml whey solution,then 20 ml of TiO₂(0.025m) were added to the first and second flask respectively and both were stirred for half hour by magnetic stirrer while the third flask contain whey only. *Lactobacillus crispatus* was cultured in first and third flask into CO₂-incubator at 37C° for (24,48,72) hours. Second flask was used as blank for first one , the change in color observed and production of sediment was observed as primary detection of produced TiO₂ nanoparticles. For another experiment 10% glucose was added to whey solution and the same procedure for TiO₂ nanoparticles synthesis was done.

Characterization of TiO₂nanoparticles

Samples of synthesized nanoparticles were characterized after 72 hours of incubation.The formation of metal oxide TiO₂ nanoparticles was confirmed by X-ray diffraction (XRD) technique, Atomic Force Microscopy(AFM) and Scanning electron microscopic(SEM).

Results and Discusion

In the present study *Lactobacillus crispatus* confirmed as positive for biosynthesis of TiO₂ nanoparticles in different media included MRS broth, JKH₁ broth and Whey solution (with and without glucose). For all samples, solution color changed from light to dark and sediment was observed.

The XRD pattern of the sample showed the presence of peaks ($2\theta=25.3, 37.9, 54$ (anatase form)). The main peak of $\theta=25.3$ (Fig.1) matches the (101) crystallographic plane of anatase of TiO₂ nanoparticles, indicating that nanoparticles structure dominantly correspond to anatase crystalline (Byranvand *et al.*,2013),which is regarded as an attributive indicator of the biologically synthesized nanoparticles TiO₂ crystallites(Kirithi *et al.*,2011). TiO₂ is preferred in anatase form because of its high photocatalyticactivity, since it has a higher potential energy of photogenerated electrons, high specific area, non-toxic, photochemically stable and relatively inexpensive(Macwan *et al.*,2011).

The Scanning electron microscope (SEM) images of the synthesized TiO₂ nanoparticles have shown spherical, oval in shape (Fig.2 A and B). Similar result of the TiO₂ nanoparticles shape was reported by using *Lactobacillus* sp.(Jha *et al.*,2009.,Azhar *et al.*,2011).

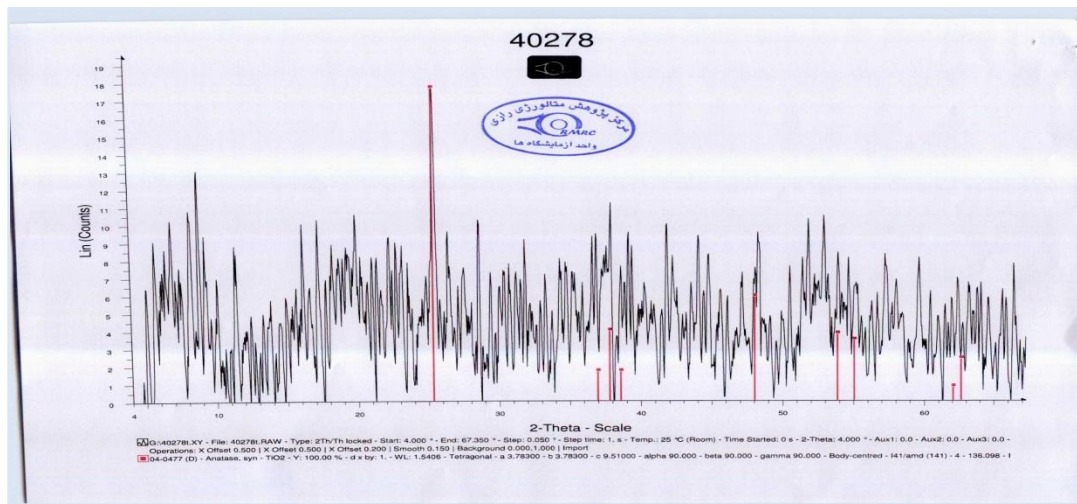


Fig.(1) :X-ray diffraction patterns of TiO₂ nanoparticles synthesized by *Lactobacillus crispatus*

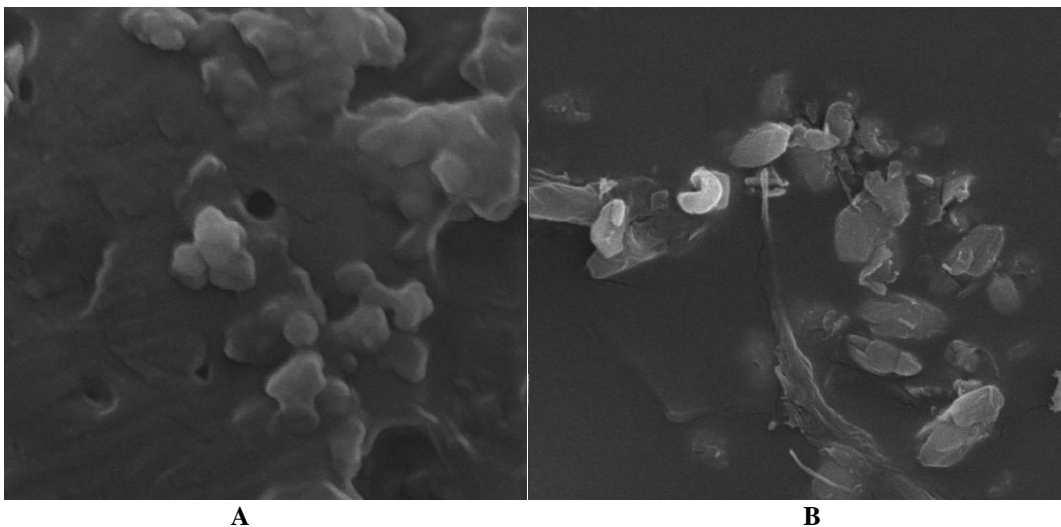
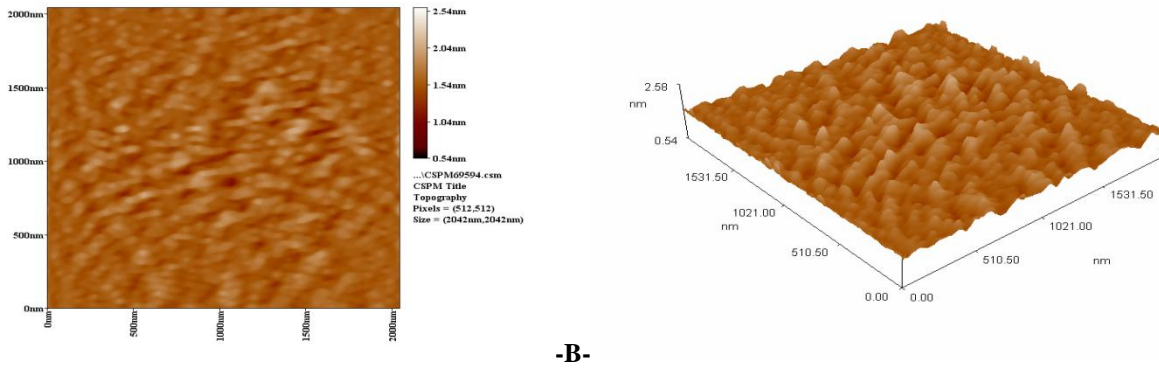
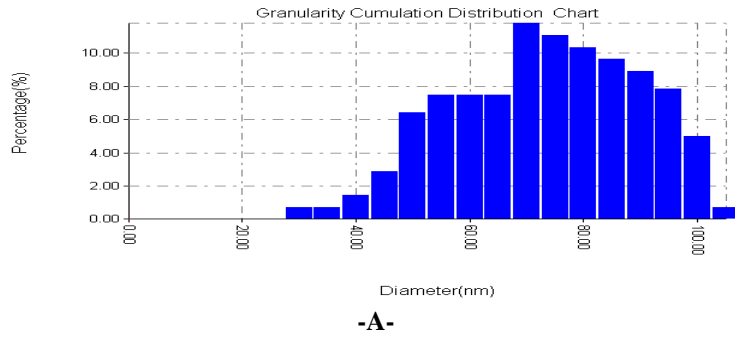


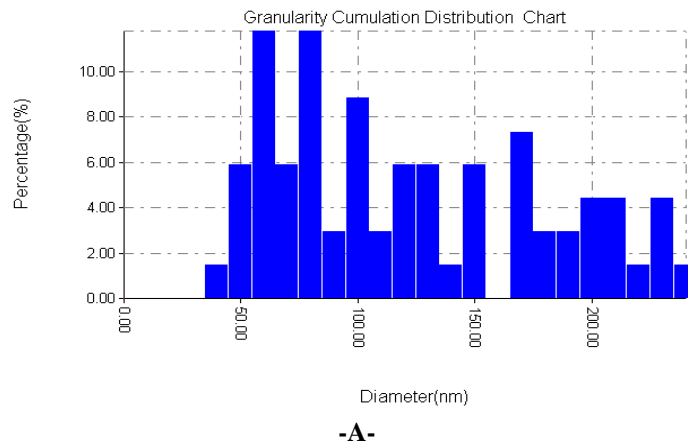
Fig.(2): (A and B) Scanning electron microscopic images of the of TiO₂ nanoparticles synthesized by *Lactobacillus crispatus*

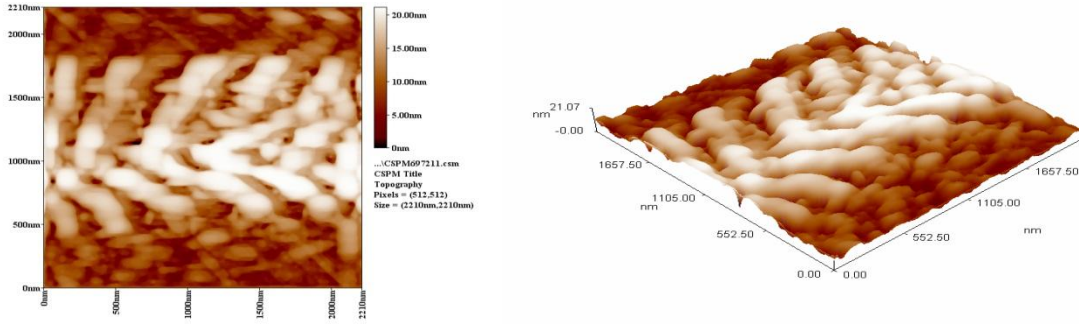
The synthesized TiO₂ nanoparticles were characterized by AFM to view the nanoparticles both in surface and three Dimensional view , and found the average size of particles (70.98 ,116.89, 113.67)nm (Fig.3,4,5) when MRS broth,JKH₁broth and whey solution were used respectively . Adding of 10% glucose to JKH₁ broth and Whey solution lead up to decreasing the average size of synthesized nanoparticles reached to (92.65 and 112.26)nm for JKH₁ broth and whey solution respectively(Fig.6,7). Jha and Prasad, (2010) showed that the lowered oxidation potential which is formed due to the presence of glucose activates the membrane bound oxidoreductase and makes the requisite ambience for an oxide nanoparticle synthesis. Azhar *et al.*,(2011) found extracellular *Lactobacillus*-mediated biosynthesis of TiO₂ nanoparticles in MRS-broth medium , with nanoparticle size 150nm. Whey solution has been used for synthesis of nanoparticles using *Lactobacillus* sp. by others studies(Prasad *et al.*,2007; Ranganath *et al.*,2012; Salman,2013). To the best of our study this is the first report on synthesis of TiO₂ nanoparticles using locally medium (JKH₁) prepared by Salman and Khalaf (2014) which considered as suitable medium for growth of *Lactobacillus* sp. and production of antibacterial agents. The present study showed that the medium JKH₁ with 10% glucose is the best for TiO₂ nanoparticles synthesis compared with the same medium without glucose.

The explanation of TiO_2 nanoparticles synthesis using *Lactobacillus* sp. is Lactobacilli, like most of the bacteria, have a negative electro-kinetic potential; which readily attracts the cations and this step probably acts as a crux of the procedure of biosynthesis (Ahmad *et al.*,2013).Extra cellular proteins and other biomolecules present in the culture of Lactobacillus mediated the hydrolysis of the anionic complexes and results in the synthesis of titanium nanoparticles (Bansal *et al.*, 2005; Jha and Prasad, 2010;Ahmad *et al.*,2013).



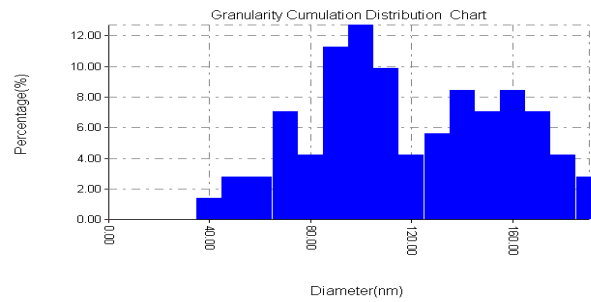
Figure(3):Atomic Force Microscopy image of TiO_2 nanoparticles synthesized by *Lactobacillus crispatus* in MRS broth.(A-Diameter percentage ,B- Surface and three Dimensional view)



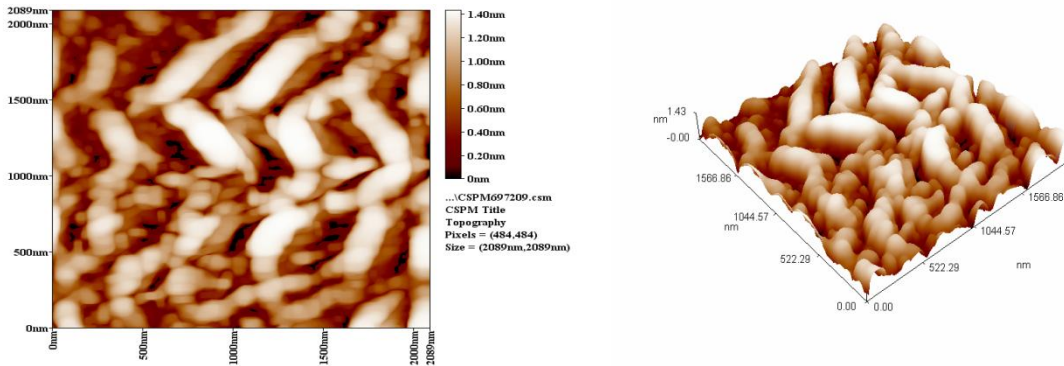


-B-

Figure(4):Atomic Force Microscopy image of TiO₂ nanoparticles synthesized by *Lactobacillus crispatus* in JKH₁ broth.(A-Diameter percentage ,B- Surface and three Dimensional view)

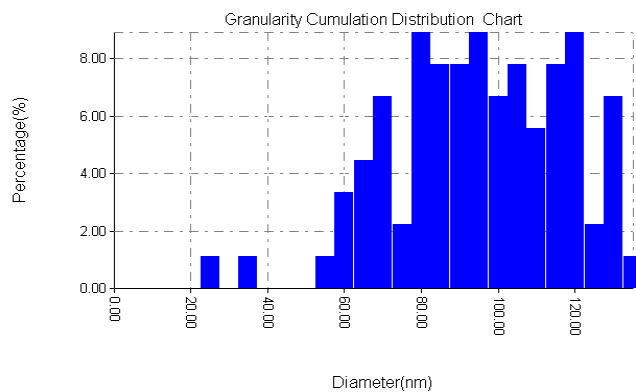


-A-

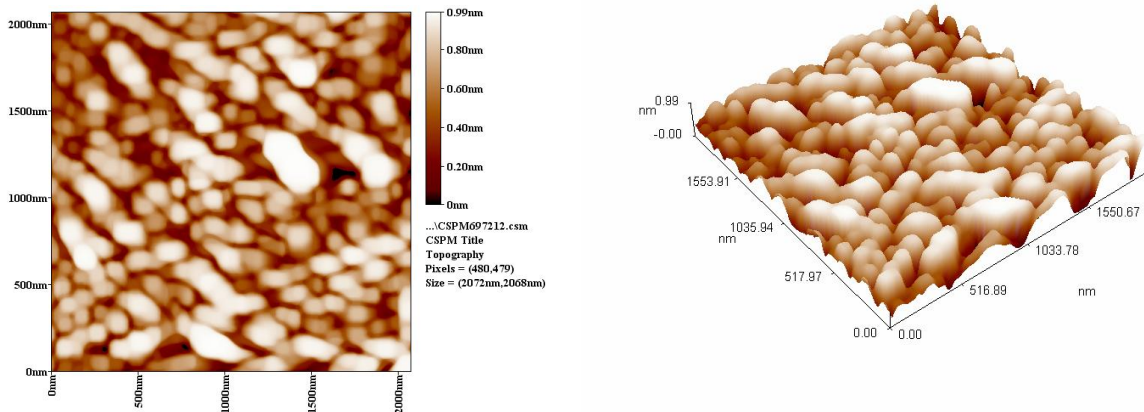


-B-

Figure(5):Atomic Force Microscopy image of TiO₂ nanoparticles synthesized by *Lactobacillus crispatus* in whey.(A-Diameter percentage ,B- Surface and three Dimensional view)

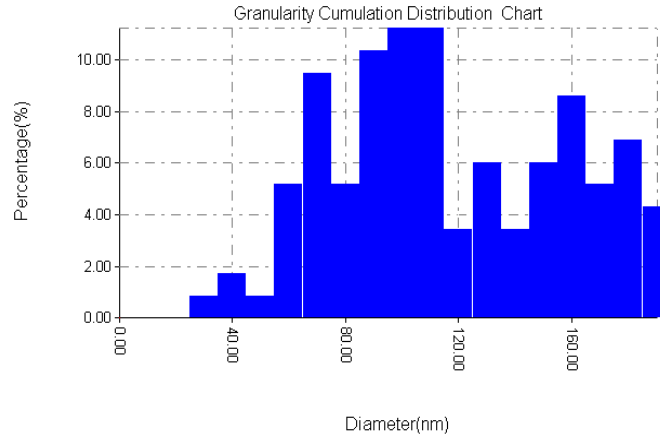


-A-

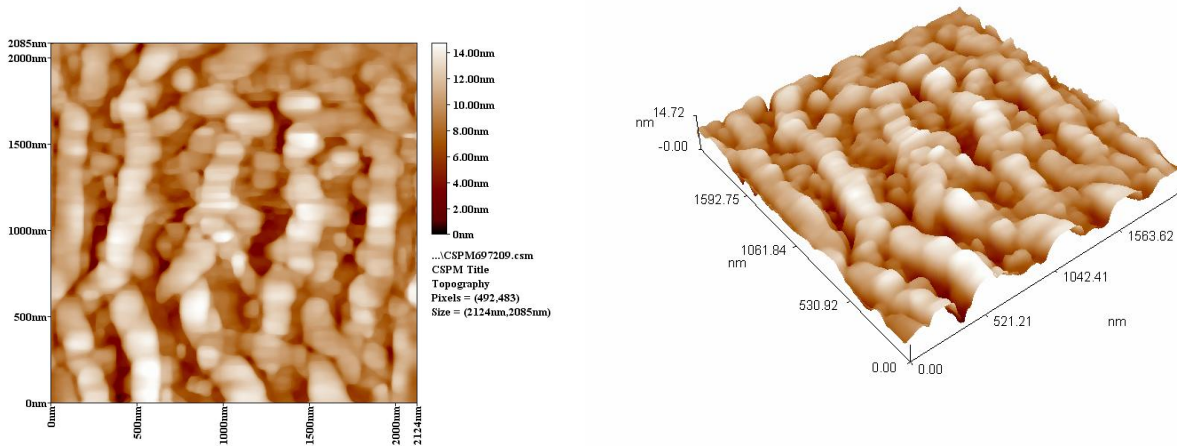


-B-

Figure(6):Atomic Force Microscopy image of TiO₂ nanoparticles synthesized by *Lactobacillus crispatus* in JKH₁ broth with glucose.(A-Diameter percentage ,B- Surface and three Dimensional view)



-A-



-B-

Figure(7):Atomic Force Microscopy image of TiO₂ nanoparticles synthesized by *Lactobacillus crispatus* in whey with glucose .(A-Diameter percentage ,B- Surface and three Dimensional view)

Conclusion:

The present study reported that the best culture media for TiO₂ nanoparticles biosynthesis using *Lactobacillus crispatus* is MRS broth followed by JKH₁ with 10% glucose .

References:

- Ahmad,R., Khatoon ,N. and Meryam ,S. (2013). Biosynthesis , characterization and application of TiO₂ nanoparticles in biocatalysis and protein folding. *Journal of proteins and proteomics.*, 4(2): 115-121.
- Azhar,A.M., Behtas Ladan, S.A., Ajabadi Ebrahimi, M.T. and Heydari,M.(2011). *Lactobacillus*-mediated biosynthesis of titanium nanoparticles in MRS broth medium . Brno, Czech Republic, EU., 9: 21– 23.
- Bansal, V., D. Rautaray, Bharde, A., Ahire, A.K. ,Sanyal, A., Ahmad, A. and Sastry, M.(2005).Fungus mediated biosynthesis of silica and titania particles. *J. Mater.Chem.*, 15:2583–2589.
- Byranvand,M.M, NematiKharat, A. , Fatholahi , L. and MalekshahiBeiranv,Z. (2013). A Review on synthesis of nano-TiO₂ via different methods. *Journal of nanostructure.*, 3: 1-9.
- Jha, A.K, Prasad , K. and Kulkarni, A.R.(2009).Synthesis of TiO₂ nanoparticles using microorganisms.*Colloids Surf B Biointerfaces.*, 71 (2):226–229.
- Jha, A.,and Prasad, K.(2010). Biosynthesis of metal and oxide nanoparticles using *Lactobacilli* from yoghurt and probiotic spore tablets.*Microbial Biotech.J.*, 5(3):285-291.
- Klaus-Joerger,T., Joerger, R. , Olsson, E. and Granqvist,G.(2001).Bacteria as workers in the living factory: metal-accumulating bacteria and their potential for materials science, *Trends Biotechnol.*, 19 (1):15-20.
- Kirthi, A.V., Abdul Rahuman , A., Rajakumar, G., Marimuthu, S., Santhoshkumar, T., Jayaseelan, C., Elango, G., AbduzZahir, A., Kamaraj, C. and Bagavan,A.(2011). Biosynthesis of titanium dioxide nanoparticles using bacterium *Bacillus subtilis*. *Materials Letters* ., 65 :2745–2747.
- Li,X., Xu, H., Chen, Z. and Chen,G. (2011). Biosynthesis of nanoparticles by microorganisms and their applications. *J. Nanomaterials* .,16.
- Macwan,D.P., Dave, P. N. and Chaturvedi,S. (2011).A review on nano-TiO₂ sol–gel type synthesis and its applications.*Mater,J. Sci.*, 46(11) :3669–3686.
- Malarkodi,C. , Chitra,K.Rajeshkumar, K., Gnanajobitha, K., Paulkumar, M., Vanaja, G. and Annadurai,G.(2013). Novel eco-friendly synthesis of titanium oxide nanoparticles by using *Planomicrobium* sp. and its antimicrobial evaluation. *Der Pharmacia Sinica.*,4(3):59-66.
- Popecu,M., Velea, A. and Lorinczi,A. (2010). Biogenic production of nanoparticles. *Digest Journal of Nanomaterials and Biostructures* .,5(4): 1035 – 1040.
- Prakash,A., Sharma, S., Ahmad , N. , Ghosh, A. and Sinha,P.(2010).Bacteria mediated extracellular synthesis of metallic nanoparticles. *International Research Journal of Biotechnology.*, 1(5) :071-079.
- Prasad , K., Anal, K. J. and Kulkarni, A. R.(2007) . *Lactobacillus* assisted synthesis of titanium nanoparticles. *Nanoscale Res Lett* ., 2:248–250.
- Ranganath,E., ,Rathod, V. and Banu,A.(2012). Screening of *Lactobacillus* spp, for mediating the biosynthesis of silver nanoparticles from silver nitrate. *Journal of Pharmacy* .,2(2) : 237-241.
- Sagadevan,S.(2013). Synthesis and electrical properties of TiO₂ nanoparticles using a wet chemical technique. *American Journal of Nanoscience and Nanotechnology.*,1(1): 27-30.
- Salman,J.A.S.(2013). Antibacterial activity of silver nanoparticles Synthesized by *Lactobacillus* spp.against Methicillin Resistant-*Staphylococcus aureus*. *International Journal of Advanced Research* .,1(6): 178-184.
- Salman,J.A.S., and Khalaf, K.H.J.(2014). Preperation of locally culture media for probiotic bacteria. *Iraqi journal of science and technology.*,(in press).
- Tripathy, A.,A., Raichur , M. , Chandrasekaran, N. , Prathna, T.C. and Mukherjee, A. (2010).Process variables in biomimetic synthesis of silver nanoparticles by aqueous extract of *Azadirachta indica* (Neem) leaves. In: *Journal of Nanoparticle Research.*,12 (1): 237-246.