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### RESEARCH ARTICLE

#### PHYSICO-CHEMICAL CHARACTERISATION OF 'UTTAMARASASINDURA': A VALUABLE KUPIPAKWARASAYANA FROM 'RASARAJACINTAMANI'

Dr. Sajina P.<sup>1</sup>, Dr. Asha P.N.<sup>2</sup> and Dr. A.K. Muraleedharan<sup>3</sup>

1. PG Scholar, Department of Rasashastra & Bhaishajyakalpana, MVR Ayurveda Medical College, Parassinikkadavu, Kannur, Kerala, India.
2. Guide and Associate Professor, Department of Rasashastra & Bhaishajyakalpana, MVR Ayurveda Medical College, Parassinikkadavu, Kannur, Kerala, India.
3. Professor & HOD, Department of Rasashastra & Bhaishajyakalpana, MVR Ayurveda Medical College, Parassinikkadavu, Kannur, Kerala, India.

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#### Abstract

Ayurveda, literally means the science of life is a comprehensive system of health care of great antiquity, based on experiential knowledge and grown with perpetual additions. Rasashastra (Iatrochemistry) is an integrated part of Ayurveda which is based on *Rasa* (Mercury) and *Rasadravyas* (Mercury related substances). Rasarajacintamani is a compiled textbook of Rasashastra written in Malayalam language by Vadayattukotta K Parameshwaran Pillai. *Uttamarasasindura* is explained in this textbook along with different types of *Rasasinduras*. The treatise Rasarajacintamani is reviewed for the specific formulation, *Uttamarasasindura*. Analytical study is needed to evaluate the proper formation and chemical composition of the medicine. Contextual reviewing revealed that *Uttamarasasindura* is a multidimensional remedy of clinical practice.

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#### Introduction:-

Rasashastra is considered as the ancient science of pharmaceuticals. The branch has discovered many formulations and well known with their quicker action in smaller doses and longer stability period. Ayurveda made a landmark in the history of medicine after the development of Rasashastra (Iatrochemistry) by judicious therapeutic administration of Herbo-Mineral preparations with high degree of safety and efficacy. In the classical textbooks of Rasashastra, different forms of mercurial preparations were mentioned. *Kupipakwa Rasayana*<sup>[1]</sup> is one among them. *Kupipakwa Rasayana* or *Sindurakalpana* preparations differ due to their proportion of mercury and sulphur. The concept of *gandhakajarana*<sup>[2]</sup> (heat treatment given on purified mercury in the presence of purified

sulphur) is one of the prime concepts among the *Rasashastra Acaryas*. *Rasarajacintamani* is a compiled textbook of Rasashastra written in Malayalam language by Vadayattukotta K Parameshwaran Pillai. This book can be taken as by the virtue of its practical usefulness. *Uttamarasasindura*, also known as *askuppisindura* is explained in this textbook along with different types of *Rasasinduras* (eg: *Jyothishangarasasindura*). Administration of *Rasasinduras* with different *Anupanas* are also mentioned in this context. The conversion of '*Kajjalirupa Rasa*' (Black sulphide of mercury) to '*Sindura*' is a complex process. Thus occurred product is having action in the deeper *Dhathu* level, also capable of furnishing *Rasayana* effect. *Uttamarasasindura* contains *Navasadara* (Salammoniac) along with *Parada* (mercury)

Corresponding Author:- Dr. Sajina P.

Address:- PG Scholar, Department of Rasashastra & Bhaishajyakalpana, MVR Ayurveda Medical College, Parassinikkadavu, Kannur, Kerala, India.

and *gandhaka* (sulphur). Three different *Bhavanadravyas* (media of levigation) are also mentioned here. Thus it differs from normal *Rasasindura*. Different *Anupanas* are mentioned for this formulation and explained as *Sakalarogahara*. For pharmacological and therapeutic assessment of the compound structural and chemical analysis is a necessary part. So in the present study, *Uttamarasasindura* prepared with *Ashtasamskara Parada* was prepared and its physicochemical characterization was carried out by using sophisticated techniques such as X-ray diffraction (XRD), Fourier transform infrared spectroscopy (FTIR), Scanning Electron Microscopy (SEM) and Energy Dispersive X-Ray Analysis (EDAX).

### Materials and Methods:-

Test drug *Uttamarasasindura* (URS)<sup>[3]</sup> was prepared as per the reference from Rasarajacintamani in the departmental laboratory. Raw Material *Parada* (mercury), *Amalasar Gandhaka* (yellow sulphur) and *Navasagara* (Sal ammoniac) were purchased from local market, and drugs for levigation like *citrakamoola* (*Plumbagoindica* Linn.), *datura* (*Daturametel* Linn.) and *kumari* (*Aloe barbadensis* Mill) were collected and identified from the Pharmacy of MVR Ayurveda medical college, Parassinikadavu, Kannur, Kerala.

*Parada* was subjected to *sodhana* by methods of *Ashtasamskara*<sup>[4]</sup>. *Gandhaka* was processed to *Shodhana* by *kurmaputavidhi*<sup>[5]</sup> and *navasagara* was prepared as per the classical method of *navasagarasodhana*<sup>[6]</sup>. Processed mercury and sulphur were taken in ratio of 1:1 and triturated in a clean *khalwayantra*. After about 38 hrs of trituration prepared *navasagara* (1/2 part) was added and trituration continued until fine lusterless powder of black sulfide of mercury (*Kajjali*)<sup>[7]</sup> was obtained. *Kajjali* was levigated with decoction of *citrakamoola* (*Plumbagoindica* Linn.), juices of *datura* (*Daturametel* Linn.) and *kumari* (*Aloe barbadensis* Mill). Levigation was done for 24hrs each with each of the media (Table 01). The fine powder was filled in the specially prepared glass bottle (*LepitaKacakupi*)<sup>[8]</sup> and heated for 16 ½ hrs. The heat was provided in controlled manner and gradually increasing temperature in conventional *kupipakwa* method using *valukayantra*<sup>[9]</sup> (Diagram.01). After the desired characteristic features of the preparation was obtained, mouth of glass bottle was sealed, and sand was removed from the neck portion from outside and allowed for self-cooling. The peak temperature obtained was 672.3<sup>0</sup> C (Table 02). Sublimed final product was collected from the neck of the glass bottle [Figure 01]. It was powdered, triturated for 3 hrs and used for further analysis [Figure 02]; labelled as sample URS. Physicochemical analysis were carried out for determination of structural characterization of URS.

**Table 01:-** Showing quantity of ingredients used for preparation of URS.

Qty. of <i>Parada:Gandhaka:Navasagara</i> taken	<i>Kajjali</i> obtained	Percentage of Yield	Total qty. of <i>bhavanadravya</i> used	<i>Kajjali</i> taken for <i>bhavana</i>	Qty. after <i>bhavana</i>
96g :96g :48 g	236g	98.3 %	183 ml	234 g	235.5g

**Table 02:-** Showing details of preparation of URS.

Qty. of <i>kajjali</i> used	Duration of heat given	Peak temperature	Wt. of URS obtained	% of yield
233 g	16 ½ hrs.	672.3 <sup>0</sup> C	94.5 g	40.5 %

### Observations and Results:-

During gradual heating process of *kupipakwavidhi* different stages shown characteristic observations (table 03). The powdered URS was found tasteless, odorless, fine and bright brick red colored powder on physical examination (Table.04). Characterization by instrumental

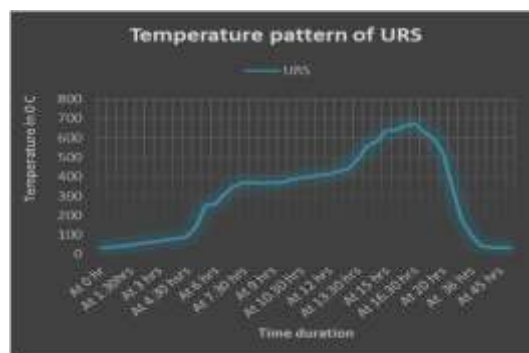
analysis<sup>[10]</sup> such as X-ray diffraction (XRD), Fourier transform infrared spectroscopy (FTIR), Scanning Electron Microscopy (SEM) and Energy Dispersive X-Ray Analysis (EDAX) were carried out.

**Table 03:-** showing observations during URS preparation.

Time duration	Observations
At 4 hrs	Pale-white fumes inside the kupi; <i>kajjali</i> remained in powder form.;Characteristic smell of <i>navasadara</i> arrived
At 4.35 hrs	Pale-white fumes came out
After 5 hrs	Characteristic odour of <i>gandhaka</i> started;melting of <i>gandhaka</i> started.confirmed with <i>seetasalaka</i> .
At 5.50 hrs	Dense white fumes
At 6.10hrs	Yellow coloured fumes with typical rotten egg smell
At 6.30 hrs	Mouth of <i>kupi</i> blocked with more <i>gandhaka</i> ;small flames appeared on <i>taptasalaka</i> insertion
At 8 hrs	More blockage and height of the flame increased
At 8.15 hrs	Continuos bluish flame from the mouth of <i>kupi</i>
At 9 <sup>th</sup> hr	Orange-brown deposits at the neck and mouth region with glittering appearance
At 10 <sup>th</sup> hr	Boiling <i>kajjali</i> found at the base with an orange shade
At 10.45 hrs	More Flames and dense yellow fumes on <i>tapta salaka</i> insertion
At 13 <sup>th</sup> hr	Golden coloured glittering at the neck region
At 15 <sup>th</sup> hr	Appearance of <i>balaruna varna</i> (bright red colour of rising sun)
At 16.30 hrs	Fumes and flames Completely stopped,confirmed with copper coin test and corking was done.

**Table .04:-**showing physico- chemical parameters <sup>[11]</sup> of URS

Parameter	Results
Colour obtained	Brick red colour( <i>sinduravarna</i> )
pH value	7.3
Total ash/Ash value	10.11 ± 0.08
Acid insoluble ash	0.48 ± 0.10
Water solubleash	1.20 ±0.25
Loss On Drying	7.16 ±0.12

**Diagram 01** showing temperature pattern of URS**XRD analysis**

XRD analysis was done using the instrument Rigaku Miniflex 600 X-ray diffractometer by using X-ray wavelength, CuK alpha -1.54056 Angstroms and with a power 30kv and 15mA. Sample was scanned from 5 degree to 90 degrees (2θ value). Crystallite size (D) is calculated by using Scherer's formula. Red cinnabar (α-

HgS), Crystal Structure: Hexagonal, Space group: P3221 is identified by the standard data JCPDS no: 01-080-2192.[table 05,table 06,diagram 02].

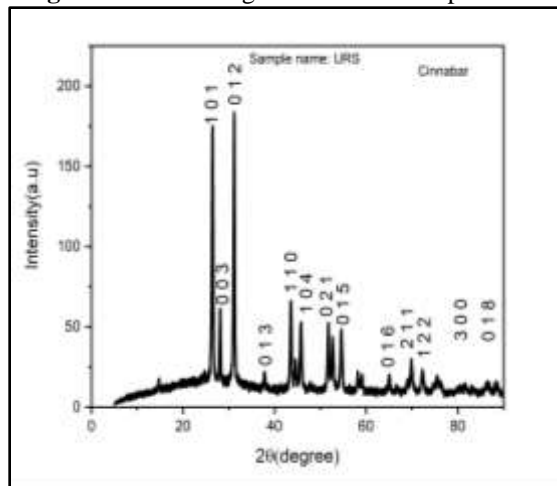
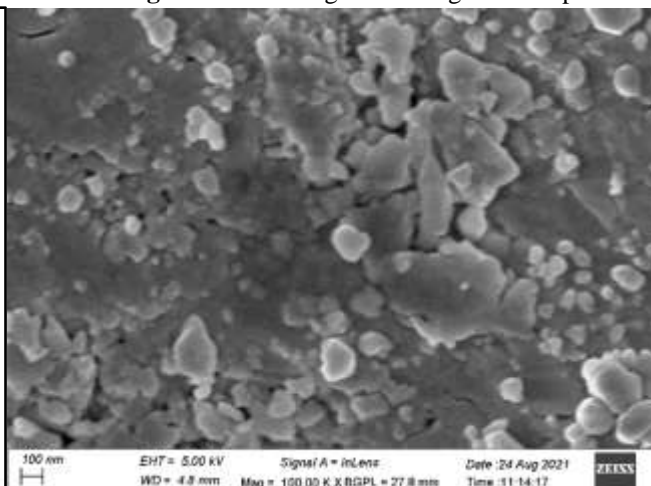
Peaks corresponding to (101),(003),(102),(013),(110),(104),(021),(015),(016),(211),(122),(300),(018) planes are obtained for HgS.

**Table 05:-**showing observations and results of XRD analysis for sample URS.

2θ (degree)	d value (Å <sup>0</sup> )	Intensity (cps)	Crystallite Size (Å <sup>0</sup> )
26.45	3.36678	942.26	260
28.24	3.15792	312.74	224
31.16	2.86844	858.96	314
37.69	2.38461	128.04	230
43.57	2.07526	316.37	252
44.62	2.02911	89.03	234
45.78	1.97998	253.59	236
51.73	1.76561	238.13	283
52.72	1.73467	207.36	238
54.53	1.68154	247.67	221
58.14	1.58526	97.11	188
59.31	1.55663	105.98	250
64.95	1.43457	101.23	258
69.88	1.34486	181.19	306
72.28	1.30606	84.98	209
75.41	1.25943	91.27	275

**Table 06:-** Showing identified compound details of sample URS.

Sample Name	Compound name	Formula	Concentration level	Crystal Structure
URS	Cinnabar	HgS	Major	Hexagonal

**Diagram 02:-** showing XRD data of sample URS**Figure 03:-** showing SEM-images of sample URS**SEM Analysis**

The SEM analysis reveals that *Uttamarasasindura* particle size lies in micro-nano sized range and has an

average size of 100 nm at 100.00Kx.[Table 07,figure 03]

**Table 07:-** Showing SEM analysis results of URS sample.

URS	Magnification	Particle size.	Magnification	Particle size.
Area 1	50.00K x	200nm	2.50Kx	10 μm
	25.00Kx	1 μm	1.00 Kx	10 μm
	10.00Kx	1 μm	500 x	20 μm
	5.00 Kx	2 μm		
Area 2	100.00 Kx	200 nm	5.00 Kx	2 μm
	75.00 Kx	200 nm	2.50 Kx	10 μm
	50.00 Kx	200 nm	1.00 Kx	10 μm
	25.00 Kx	1 μm	500 x	20 μm
	10.00 Kx	1 μm		
Area 3	100.00 Kx	100 nm	5.00 Kx	2 μm
	50.00 Kx	200 nm	2.50 Kx	10 μm
	25.00 Kx	1 μm	1.00 Kx	10 μm
	10.00 Kx	1 μm	500 x	20 μm

**EDAX Analysis**

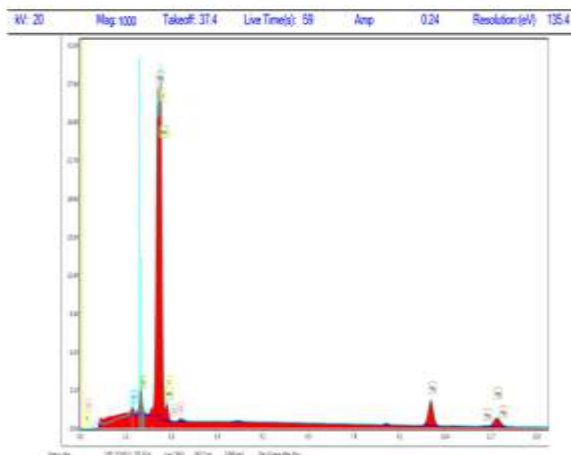
EDAX analysis was carried out in association with SEM. Data shows Mercury and sulphur as the main contents.[Table08,Table09,Table 10,Diagram 03].

**FT-IR Spectroscopy**

Infrared Qualitative Analysis was performed using NICOLET iS50 FT-IR. Used test method was based on

ASTM E1252-98(Re-approved 2021) and E573-01(Re-approved 2021).KBr pellet was used for background correction.Sample was mixed with KBr, pelletized and analysed in transmission mode.In the FTIR spectrum of sample URS two peaks are obtained at  $3447\text{cm}^{-1}$  and  $1070\text{cm}^{-1}$ .These peaks can be assigned to O-H stretching and O-H deformation.[Diagram 04]

**Diagram 03:-** showing EDAX result of URS sample



**Table08:-**showing results of EDAX-Areal

Element	Weight %	Atomic %	Net Int.
Al	0.73	2.83	93.13
Si	0.65	2.41	101.01
S	15.92	51.68	2036.37
Cl	0.08	0.23	5.85
Hg	82.61	42.86	674.49

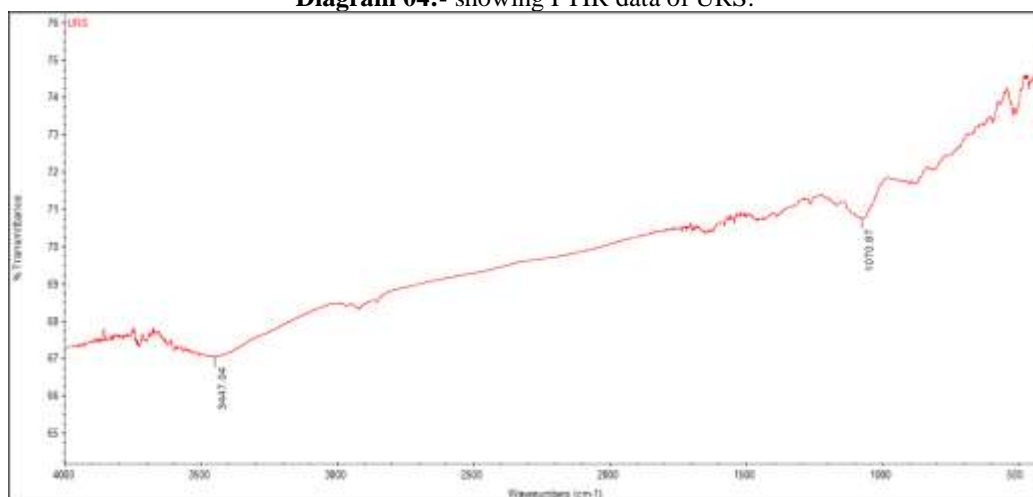
**Table 09-** showing results of EDAX-Area 2

Element	Weight %	Atomic %	Net Int.
Al	1.13	4.11	153.1
Si	0.92	3.2	150.8
Mo	1.96	2.01	131.12
S	16.9	51.67	2293.32
Cl	0.17	0.47	13.49
Hg	78.92	38.55	674.41

**Table 10:-** showing results of EDAX-Area 3

Element	Weight %	Atomic %	Net Int.
Al	1.36	4.89	178.31
Si	0.92	3.17	145.86
Mo	1.61	1.63	103.85
S	16.94	51.25	2219.37
Cl	0.34	0.94	26.47
Hg	78.82	38.11	650.22

**Diagram 04:-** showing FTIR data of URS.



**Discussion:-**

In conceptual study, *Uttamarasasindura* has been described as very effective and as a wide angle remedy. On qualitative test, the *sindura* sample (URS)

showed relevant values. Test for acid insoluble ash was carried out to evaluate the percentage of insoluble inorganic content of the *sindura* in dilute acid. Since a drug must first pass into solution before it can be

absorbed, the acid insoluble ash test is therapeutically very important. It is intended to provide a step towards the evaluation of the physiological availability of the product. The obtained value is  $0.48 \pm 0.10$  which signifies the genuinity of the product. The water soluble ash value was found as  $1.20 \pm 0.25$ .

The total ash value has importance in *sindura* preparation because total ash may act as source of trace elements. Physico chemical changes of media during *Shodhana*, *bhavana* may contribute to this. It is calculated as  $10.11 \pm 0.08$ . pH was calculated as 7.3 which is acceptable to body tissues. LOD for the product was carried out. This test is to detect the moisture and volatile content in the sample. The analysed value  $7.16 \pm 0.12$  is indicating stability and more shelf life of *Uttamarasasindura*.

SEM analysis of URS shows surface is ultra smooth with uniform morphology as shown in the result part. The particle size ranged from 100nm to several  $\mu\text{m}$ . The EDAX shows the chemical composition consist of mercury and sulphur. As the particle size was in nanometer, absorption of the drug will be more with quicker action. EDAX refers to energy dispersion X-ray spectroscopy and is employed here as an analysis tool to determine the elemental composition of *Rasoushadhis*. It works by analyzing the spectrum of emitted X-Rays from a sample as a beam of high energy electron is incident upon it. Here EDAX worked in association with SEM. EDAX gives weight % as well as atomic % of the major elements. Selected areas of URS sample were tested. Two elements were observed in major concentration in the tested sample. Observations in Area 1, Area 2 and Area 3 revealed Hg with percentage weight of 82.61%, 78.92% and 78.82% and atomic% of 42.86%, 38.55%, 38.11% respectively. Sulphur (S) with percentage weight of 15.92%, 16.9%, 16.94% and atomic% of 51.68%, 51.67%, 51.25% respectively. The atomic% of Hg is less but its mass% is more it is because Hg is heavy metal. Other major elements are Al, Si, Mo and Cl. During *citrakasodhana* the *rakatacitraka* roots were kept immersed in an aluminium vessel. That may be the reason for presence of aluminium in the sample. Area 2 and area 3 contains Mo. Other mentioned elements are similar in all the three selected areas with slight variations in their values. The element Cl may be a contribution from the *bhavanadravyakumari*. Because the ion content of leaves and sprouts of aloe vera contains Cl. *Drugnavasada* also can be a source of Cl. FTIR analysis was carried out for samples of *bhavitakajjali* (BKJ) and the final product *Uttamarasasindura* (URS). The presence of functional group is very much essential in drug absorption as it increase the rate of absorption, on the other hand

multiple functional groups may hinder drug absorption by hindering the action of other functional group. In the FTIR spectrum of Sample BKJ-two peaks are obtained at  $3148 \text{ cm}^{-1}$  and  $1401 \text{ cm}^{-1}$ . Peak at  $3148 \text{ cm}^{-1}$  can be assigned to N-H bond stretching vibrations. It can be due to presence of ammonium salt or amine formation. Peak at  $1401 \text{ cm}^{-1}$  can be assigned to O-H deformation or C-O stretching. Presence of peaks together can be considered as an evidence for the formation of ammonium salt of carboxylic acid. In the FTIR spectrum of sample URS, two peaks are obtained at  $3447 \text{ cm}^{-1}$  and  $1070 \text{ cm}^{-1}$ . These peaks can be assigned to O-H stretching and O-H deformation. In the conversion of BKJ to URS, a reduction in number of functional groups has been observed, which in turn reveals the quality of the finally formed *Uttamarasasindura*. The XRD analysis of the samples *kajjali* (KJ), *bhavitakajjali* (BKJ) and *Uttamarasasindura* (URS) showed presence of HgS (cinnabar) in major concentration level. Crystal structure is observed as hexagonal. More than 15 identical peaks of HgS were observed in the final URS sample. KJ and BKJ samples showed peaks of concentration of Chlorine with an orthorhombic crystal structure. About 13 peaks were observed in KJ, BKJ samples. From this results it is clear that the final product formed (URS) is in pure HgS form. Peaks corresponding to (101), (003), (102), (013), (110), (104), (021), (015), (016), (211), (122), (300), (018) planes are obtained for HgS.

### Conclusion:-

*Uttamarasasindura* is a tasteless, odorless, fine and bright brick red colored substance. Chemical characterisation through various techniques shows the final product formed (URS) is in pure HgS form. It contains different functional groups. The particle size of the URS ranged from 100nm to several  $\mu\text{m}$ . *Uttamarasasindura* has been described as very effective and as a wide angle remedy in the text 'Rasarajacitamani', a book with lot stress on practical applications.

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### Conflicts of interest

There are no conflicts of interest.

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encouragement throughout the work.I would like to pay high regards to my parents ,other family members and my classmates for their sincere encouragement and inspiration throughout.

**Figure 01**-Sublimed final product



**Figure 02**-Powdered sample URS



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